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Multi-Scale Simulation for Contaminants Transport in Organic Enriched Water Infiltration Process

Thesis Booklet

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1. Introduction

Soil aquifer treatment (SAT) systems are emerging as effective solutions for wastewater treatment. They offer environmentally sustainable and cost-effective approaches to reduce pollutant levels in water resources. This study compares different soil types in SAT systems, examines infiltration rates, wetting and drying cycles, and pollutant interactions. It sheds light on how pollutants move through SAT systems. By using experiments, models, and statistics, it can provide new insights to improve SAT system design and operation for better pollutant removal.

2. Material and methods

Comprehensive investigations used to explore contaminants behavior in SAT systems under different aspects from filtrate media of soils, infiltration rate and operating conditions. Synthetic wastewater designed and prepared in the lab mirroring the real wastewater pollutant compositions as shown in table 1, also lab-scale columns were used to simulate real-world conditions and maintain controlled experiments.

Table 1The chemical composition for the synthetic wastewater effluents

Chemicals	Mass (mg/1L)	Chemicals	Mass (mg/1L)
CH₃COONa.3H₂O	130	H ₃ BO ₃	2.65
NH₄CI	100	CuSO ₄ .5H ₂ O	15
KH₂PO₄	175	KI	15
MgSO ₄ ·7H ₂ O	100	MnSO₄.4H₂O	15
CaCl₂·2H₂O	15	(NH ₄)6Mo ₇ O ₂₄ .4H ₂ O	5
NaHCO₃	200	ZnSO ₄ .7H ₂ O	15
KCI	35	CoCl ₂ .6 H ₂ O	15
EDTA	500	Yeast extract	200
FeCl ₃ .6H ₂ O	15	Peptone	200

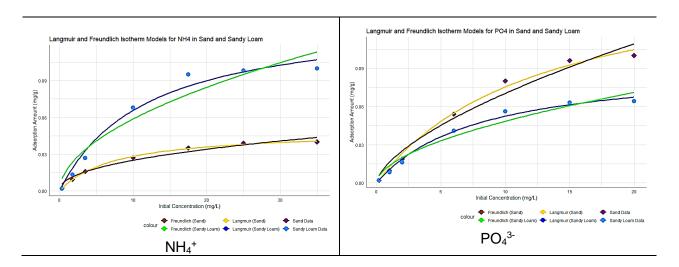
The methodology included selecting parameters such as filtrate media (Sand and Sandy loam) soils, infiltration rates (0.50 mL/min, and 10 mL/min), and operational conditions (wet and dry conditions) to provide a comprehensive and detailed understanding of how varying these conditions impact contaminants removal and transport processes from synthetic wastewater effluents through soil aquifer treatment system.

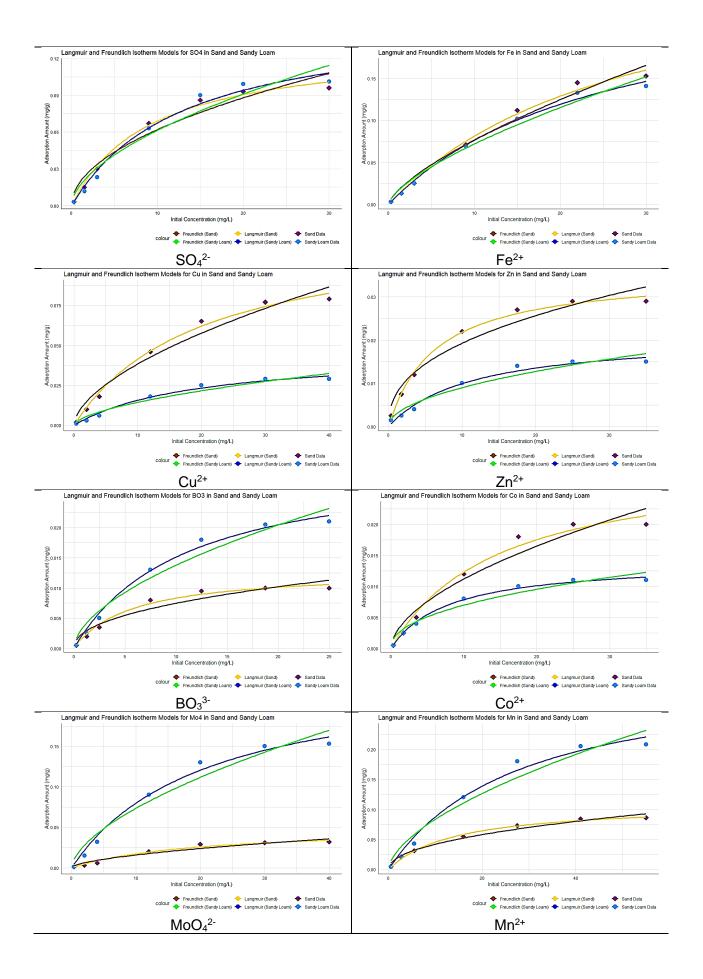
3. Comparative Investigation and Performance Evaluation of Different Soil Types in Enhancing Pollutants Removal from Synthetic Wastewater in Soil Aquifer Treatment System

This chapter present comprehensive analysis, combining static adsorption studies and dynamic column experiments, offers a novel and detailed understanding of the performance of SAT systems using sand and sandy loam soils for significantly contribute to the understanding of how different soil types perform in SAT systems and offer practical insights for optimizing the design and operation of these systems based on specific environmental conditions and pollutant profiles. The ability to correlate the adsorption capacity with real-world pollutant retention and removal efficiency provides a robust framework for designing and improving SAT systems to achieve more effective wastewater treatment.

The comparative use of Freundlich and Langmuir isothermal models provided new insights into the adsorption characteristics of sand and sandy loam soils as shown in figure 1. The findings revealed that sandy loam generally has a higher adsorption capacity for certain pollutants, while sand excels in others. The preferential fit of the Langmuir model for sandy loam pollutants further emphasizes its effectiveness in characterizing adsorption behavior, contributing to the optimization of SAT systems based on soil type.

Figure 1Freundlich and Langmuir isothermal models for synthetic wastewater pollutants in sand and sandy loam soils





The dynamic column experiments revealed significant differences in the removal efficiencies of various pollutants between sand and sandy loam soils. Sandy loam demonstrated superior removal capabilities for pollutants such as NH₄+, PO₄³⁻, Cu²⁺, Mn²⁺, MoO₄²⁻, Zn²⁺, Co²⁺, and BO₃³⁻, with removal rates substantially higher than those observed in sand. Conversely, sand exhibited greater efficiency in removing NO₂-, NO₃-, SO₄²⁻, and Fe²⁺. This differential behavior underscores the need for a tailored approach in selecting soil media based on the specific pollutants present in SAT systems. Moreover, the application of breakthrough curve analysis to quantify the area under the curve (AUC) provided a comprehensive measure of the total pollutant removal. This approach allowed for the detailed assessment of the retention capacities and adsorption behaviors of the soils, revealing the intricate processes governing pollutant removal in SAT systems. The AUC calculations highlighted the dynamic performance of sand and sandy loam soils in real-world scenarios, demonstrating the utility of this method in evaluating and optimizing SAT system designs.

4. Understanding the Role of Infiltration Rate on Contaminants Fate and Transport in Soil Aquifer Treatment System: Experimental, Modeling, and Statistical Analysis Approaches

This chapter elucidates the role of infiltration rate in pollutant removal within SAT systems. It combines experimental data, modeling insights, and statistical analysis to present a comprehensive understanding of how varying infiltration rates impact pollutant behavior and removal efficiency.

Infiltration rate significantly affects the transport and removal of various pollutants. A lower infiltration rate (0.5 mL/min) enhances pollutant removal through increased contact time with the sand media, leading to higher adsorption and more efficient biogeochemical reactions. In contrast, a higher infiltration rate (10 mL/min) reduces contact time, resulting in less effective pollutant removal and higher effluent concentrations. By identifying how different pollutants respond to changes in infiltration rate. NH₄+, PO₄³⁻, SO₄²⁻, Fe²⁺, Mn²⁺, MoO₄²⁻, Zn²⁺, and Co²⁺ these pollutants show significantly better removal at the lower infiltration rate. However, NO₃-, NO₂-, Cu²⁺, and BO₃³⁻ these pollutants exhibit higher removal at the higher infiltration rate, suggesting that specific conditions may benefit their removal.

The detailed experimental breakthrough curves demonstrate clear trends in pollutant removal efficiency under different infiltration rates. The findings confirm that a slower infiltration rate generally leads to better removal due to extended contact time.

The 2-D spatial concentration distributions and breakthrough curve modeling support the experimental observations. Models effectively capture pollutant behavior over time and depth, reinforcing the experimental results and providing a visual representation of pollutant transport dynamics.

The statistical analysis using t-tests provides quantitative evidence of the impact of infiltration rate on pollutant removal. The p-values reveal statistically significant differences in pollutant concentrations between the two infiltration rates, validating the experimental and modeling results.

The chapter findings highlight the importance of customizing infiltration rates based on specific pollutants to optimize removal efficiency. This approach allows for tailored SAT system operation, potentially improving treatment performance for diverse wastewater compositions.

5. Quantifying the influence of dynamic pulsed of wetting and drying on the transformation and fate of synthetic wastewater pollutants in soil aquifer treatment system

This chapter explains the complex interactions between wetting and drying cycles and pollutant dynamics in SAT systems. The findings contribute to a deeper understanding of pollutant removal processes and offer valuable insights for optimizing system design and operation.

The impact of wetting and drying cycles on the overall performance of the SAT system as a nature cyclic based on seasonal variation, these conditions leads to fluctuations in pollutant removal efficiency, affecting treatment performance and the physical properties of the sand column.

The use of robust statistical analysis, including logistic regression models, correlation studies and factorial interaction trends, provides a detailed understanding of the relationships between pollutants and environmental factors. This approach offers new perspectives on how these factors collectively impact pollutant behavior and treatment efficacy.

pH, Dissolved Oxygen (DO), and Oxidation-Reduction Potential (ORP) factors interact with wetting and drying cycles to influence pollutant removal. Strong correlations and interactions between these factors and pollutant concentrations highlight their pivotal role in the treatment process.

NH₄⁺ concentrations increase during wet periods due to enhanced leaching but are more effectively removed during dry periods through adsorption and potential microbial

processes. This highlights the critical role of drying phases in improving NH₄⁺ removal efficiency.

NO₂⁻ concentrations are lower during wet periods due to limited nitrification, whereas NO₃⁻ concentrations show varied responses influenced by DO, pH, and ORP. This underscores the differential behavior of nitrogen species under varying wetting conditions.

Variability in SO₄²⁻ concentrations with respect to wet and dry cycles and their association with environmental factors such as pH and ORP provides new insights into the complexities of sulfate dynamics in SAT systems.

Increased PO₄³⁻ concentrations with higher DO, pH, and ORP, and more effective removal during dry periods, illustrate the importance of environmental conditions in phosphate management.

Fe²⁺, Cu²⁺, Mn²⁺, MoO₄²⁻, Zn²⁺, and Co²⁺ specific patterns in the transport and removal of these metal ions, with wet conditions generally favoring lower concentrations. Each metal ion exhibits distinct correlations with DO, ORP, and pH, indicating their unique responses to environmental conditions.

A unique flocculation pattern in BO_3^{3-} concentrations, with higher levels during wet periods due to leaching and dissolution. This finding suggests complex interactions between BO_3^{3-} adsorption, precipitation, and desorption processes.

In summary, wetting and drying cycles emerge as critical determinants in the fate, transport, and treatment of infiltrated wastewater pollutants in the sand column. Recognizing and incorporating these cycles into system design and operation protocols is paramount for achieving and maintaining optimal system efficiency.

6. Conclusion

This thesis has advanced our understanding of soil aquifer treatment (SAT) systems and their efficacy in removing pollutants from wastewater. Through comprehensive investigations into the static adsorption behavior, dynamic breakthrough curve analysis, influence of infiltration rates, wetting and drying cycles, and synergistic and competitive pollutant interactions, key insights have been gleaned to inform system optimization strategies. The integration of experimental, modeling, and statistical approaches has facilitated a comprehensive assessment of pollutant behavior in SAT systems, highlighting the importance of tailored design considerations and management practices. Moving forward, these findings will contribute to the development of sustainable wastewater treatment solutions, safeguarding sustainable water resources.

7. New Scientific Results

Thesis 1. Comparative soil types in pollutant removal for SAT systems

I used two methods in comparing the performance of sand and sandy loam soils in removing pollutants, while also assessing the scalability of the results based on SAT system considerations. The first method involved the isothermal adsorption technique, which was used to study the static and equilibrium conditions between infiltrated pollutants and the soil particle matrix. The second method simulated real-world SAT systems through dynamic column investigations. All laboratory investigations aimed to quantify the effect of the infiltrate media and soil selection on SAT system performance.

The comparative analysis of soil adsorption performance using Freundlich and Langmuir isothermal models provided new insights in to the adsorption processes relevant to SAT systems, particularly regarding the adsorption characteristics and pollutant affinities of sand and sandy loam soils. Freundlich accommodates heterogeneous surfaces, while Langmuir provides insights into monolayer adsorption, covering a broad spectrum of adsorption behaviors. The findings revealed that sandy loam generally has a higher adsorption capacity for pollutants due to its higher organic matter content and greater pollutant affinity intensity.

Dynamic column experiments simulating real-world scenarios of partially treated wastewater infiltration revealed significant differences in the removal efficiencies of various pollutants between sand and sandy loam soils. These differences are attributed to variations in soil characteristics such as texture, structure, porosity, pH, and organic matter content. Sandy loam demonstrated superior removal capabilities for pollutants such as NH₄⁺, PO₄³⁻, Cu²⁺, Mn²⁺, MoO₄²⁻, Zn²⁺, Co²⁺, and BO₃³⁻, with removal rates substantially higher than those observed in sand. However, sand exhibited greater efficiency in removing NO₂⁻, NO₃⁻, SO₄²⁻, and Fe²⁺.

The area under breakthrough curve (AUC) calculations highlighted the dynamic pollutant removal performance of sand and sandy loam soils in real-world scenarios. This method demonstrates its utility in evaluating the performance of SAT systems, providing a practical measurable index for assessing SAT performance based on specific soil types and pollutant profiles.

Thesis 2. Infiltration rate impact on pollutants transport in SAT systems

The infiltration rate significantly influences the transport, retention and removal of various pollutants. A lower infiltration rate (0.5 mL/min) enhances pollutant removal by increasing contact time with the sand medium particles matrix, facilitating higher adsorption

and more efficient biogeochemical reactions due to prolonged interaction between pollutants and the microbial community within the system. In contrast, a higher infiltration rate (10 mL/min) reduces contact time, resulting in less effective pollutant removal and higher effluent concentrations.

Understanding how different pollutants respond to changes in infiltration rate reveals significant variations in their removal efficiencies. Pollutants such as NH₄+, PO₄³⁻, SO₄²⁻, Fe²⁺, Mn²⁺, MoO₄²⁻, Zn²⁺, Cu²⁺ and Co²⁺ exhibit significantly better removal at the slower infiltration rate due to enhanced adsorption and microbial degradation interactions. However, pollutants such as NO₃-, NO₂-, and BO₃³⁻ show higher removal at the higher infiltration rate likely due to reduced production rates, particularly for nitrogen species. These findings suggest that specific conditions tailored to the pollutant profile may optimize removal processes.

The two-dimensional spatial concentration distributions and breakthrough curve modeling, developed using random walk technique, represent an innovative approach that effectively simulates particle dispersion and advection in porous media. This method demonstrates high efficiency and accuracy, particularly under conditions with steep pollutant gradients, making it a valuable tool for modeling dynamic contaminant transport plumes in SAT systems. The developed models calibrated with experimental observations by accurately capturing pollutant behavior over time and depth. These models reinforce experimental results and provide a visual representation of pollutant transport dynamics while characterizing key transport parameters. Capillary diffusivity, which drives solute movement in unsaturated soils, surpasses molecular diffusion and dispersion in dominance, especially under low infiltration rates where it enhances adsorption and retention.

The statistical analysis using t-tests provides quantitative evidence of the impact of infiltration rate on pollutant removal. The p-values reveal statistically significant differences in pollutant concentrations between the two infiltration rates, validating the experimental and modeling results.

Thesis 3. Impact of pulsed wetting and drying on pollutant fate in SAT systems

The impact of wetting and drying cycles on the overall performance of the SAT system as a nature cyclic based on seasonal and operational variation, these conditions leads to fluctuations in pollutant removal efficiency, affecting treatment performance and the physical properties of the filtrate media.

The use of robust statistical analysis, including logistic regression models, correlation trends and factorial interaction analysis as an innovation framework to validate

and track the geochemical interactions, provides a detailed understanding of the relationships between pollutants and climatical and operational factors. This approach offers new perspectives on how these factors collectively impact pollutant behavior and treatment efficacy.

pH, Dissolved Oxygen (DO), and Oxidation-Reduction Potential (ORP) factors interact with wetting and drying cycles to influence pollutant removal. Strong correlations and interactions between these factors and pollutant concentrations highlight their pivotal role in the treatment process.

Nitrogen species (NH₄⁺, NO₂⁻, and NO₃⁻) exhibit dynamic behaviors during wet and dry periods in the SAT system. NH₄⁺ concentrations increase during wet periods due to enhanced leaching but are more effectively removed during dry phases through adsorption and nitrification potential microbial processes, highlighting the critical role of drying in improving NH₄⁺ removal efficiency. In contrast, NO₂⁻ concentrations are lower during wet periods due to limited nitrification, while NO₃⁻ concentrations show varied responses influenced by DO, pH, and ORP, underscoring the differential behavior of nitrogen species under fluctuating wetting conditions.

Increased PO₄³⁻ concentrations during wet periods, associated with higher DO, pH, and ORP interactions, and more effective removal during dry periods due to immobilization and geochemical reactions, highlight the importance of wet/dry conditions in phosphate management. Meanwhile, The variability in SO₄²⁻ concentrations during wetting and drying cycles, influenced by wet/dry factors such as pH and ORP, provides new insights into the complexities of sulfate dynamics in SAT systems.

Fe²⁺, Cu²⁺, Mn²⁺, MoO₄²⁻, Zn²⁺, and Co²⁺ specific patterns in the transport and removal of these metal ions, with wet conditions generally favoring lower concentrations. Each metal ion exhibits distinct correlations with DO, ORP, and pH influencing the oxidation or precipitation of these metals, indicating their unique responses to wet/dry conditions. In contrast, BO₃³⁻ concentrations display a distinct fluctuation pattern, with higher levels during wet periods due to leaching and dissolution. This finding suggests complex interactions between BO₃³⁻ adsorption, precipitation, and desorption processes.

8. Author Individual Scientific Works

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